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~~Energy Potential Energy = position energy (chemical bonds) Kinetic Energy = motion energy (heat) Thermochem 3 First Law of Thermodynamics. First Law is conservation of energy, or Energy of universe is constant. Heat flow exothermic (heat out of system) (-) and endothermic (heat into system) (+)~~

Thermochemistry

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162 CHAPTER 6: THERMOCHEMISTRY To convert the answer to joules, we write: 101.3 J 0.18 L atm = - x = - w - 18 J 6.17 An expansion implies an increase in volume, therefore w must be - 325 J (see the defining equation for pressure-volume work.)

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184 Guided Reading and Study Workbook CHAPTER 17, Thermochemistry (continued) 6. In thermochemical calculations, is the direction of heat flow given from the point of view of the system, or of the surroundings? 7. What universal law states that energy can neither be created nor destroyed

SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505–510)

You must be able to explain different types of reactions and know how to write thermochemical equations to answer all questions on this quiz. Quiz & Worksheet Goals.

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