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the pH of a 0.250 mol/L solution  
of HCl(aq). What is pOH of the  
same solution? 25. What is the pH  
of a solution produced by  
dissolving 2.75 g of HBr(g) in

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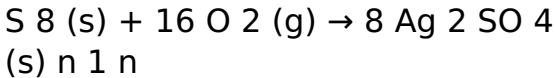
water to make 700 mL of solution? 26. A solution has  $\text{pH} = 3.9$ . Calculate  $[\text{H}_3\text{O}^+]$  and  $[\text{OH}^-]$ . 27. A solution has  $\text{pH} = 2.15$ .

Solutions Review Worksheet - Ms.  
Mogck's Classroom  
Quantitative Chemistry Review

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Worksheet page 7 6. a)  
precipitate b) filtrate c)  
erlenmeyer flask 7. The  $\text{CuSO}_4$  in  
solution was the excess reagent.  
It could be poured into the  $\text{Sr}(\text{NO}_3)_2$   
without any need of rinsing  
to make sure that all of it went  
into the reaction. 8.  $16 \text{ Ag(s)} + 1$

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Solutions Review Worksheet - Ms. Redding's Science Page  
Honors Chemistry Solutions  
Worksheet - Review REVIEW  
SOLUBILITY 1. \_\_\_\_\_ is the

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substance that is dissolved. 2. \_\_\_\_\_ is the substance that does the dissolving. 3. \_\_\_\_\_ are homogeneous mixtures. 4. A (n) \_\_\_\_\_ solution has the maximum amount of solute dissolved in a given amount of solvent. 5. A(n) \_\_\_\_\_ solution can



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dissolve more solute. ...

## Honors Chemistry Solutions Worksheet - Review

Quantitative Chemistry Review

Worksheet page 7 6. a)

precipitate b) filtrate c)

erlenmeyer flask 7. The  $\text{CuSO}_4$  in

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solution was the excess reagent. It could be poured into the  $\text{Sr}(\text{NO}_3)_2$  without any need of rinsing to make sure that all of it went into the reaction.

8.  $16 \text{Ag}(s) + 1 \text{S}_8(s) + 16 \text{O}_2(g) \rightarrow 8 \text{Ag}_2\text{SO}_4(s)$

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(Stoichimetry Worksheet Number  
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Vancouver School Board

Worksheet: \*Mixtures\* and \*Solutio  
ns\*! Name: \_\_\_\_\_ \* Class: \* \_\_\_\_\_ \* 1.



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State, with a reason, whether each of the following is a homogeneous mixture (solution) or a ...

## Mixtures and Solutions Review for Test

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Answers 1) If I make a solution by  
adding 83 grams of sodium  
hydroxide to 750 mL of water...  
To solve problem 1, you need to  
have calculated for various parts

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that there are 2.08 moles of NaOH (which has a molar mass of 40 g/mol), that there are 750 grams of water (which has a density of 1 g/mL), and that there are 41.67 moles of water (which has a molar mass of 18 g/mol).

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Concentration Review Worksheet  
- mrphysics.org

Chapter 15: Standard Review  
Worksheet 1. When an ionic  
substance is dissolved in water to  
form a solution, the water plays  
an essential role in overcoming  
the strong interparticle forces in



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the ionic crystal. Water is a highly polar substance. One end of the water molecule dipole is partially negative, and the other is partially positive. 2.

Answer Key - Home - Francis  
Howell High School

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Dilutions Worksheet – Solutions 1) If I have 340 mL of a 0.5 M NaBr solution, what will the concentration be if I add 560 mL more water to it? 0.19 M (the final volume is 900 mL, set up the equation from that) 2) If I dilute 250 mL of 0.10 M lithium acetate

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solution to a volume of 750 mL,  
what will the concentration of this  
solution be?

Dilutions Worksheet - Chemistry  
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## UNIT 4- Solutions and Solubility - Ms. Gauthier

This quiz covers solution concentration problems, such as percent by mass, percent by volume, molarity and molality.

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You will need a periodic table and a calculator to complete this quiz. Select the best answer from the choices.

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